

Tabelle 1. ^1H -NMR-Signale von **3** (270 MHz, TMS als innerer Standard, δ -Werte)

	CDCl_3 , 270 MHz	+ Eu(fod) $_3$
2-H	dd 1.54	dd 1.54
3-H	dd 0.66	dd 0.67
3'-H	dd 0.75	dd 0.76
5-H	m 1.25	m 1.26
6-H	m 2.00	m 2.03
7-H	m 1.20	m 1.22
8-H	m 1.34	m 1.35
9-H	m 2.10	m 2.18
10-H	tq 5.45	t(br) 5.64
12-H	s(br) 4.45	s(br) 5.16
13-H	s(br) 1.66	s(br) 1.85
14-H	d 0.96	d 0.97
15-H	s(br) 4.63	s(br) 4.63
15'-H	s(br) 4.81	s(br) 4.81
OAc	s 2.08	s 2.73

$J(\text{Hz})$: 2,3 = 8; 2,3' = 3.5; 3,3' = 4; 7,14 = 6.5; 9,10 = 7; 10,13 = 1.

läßt sich auch die Konfiguration der trisubstituierten Doppelbindung bestimmen. Obwohl nicht alle Signale interpretierbar sind, läßt sich durch Doppelresonanzexperimente zeigen, daß ein Cyclopropanring vorhanden sein muß, der trisubstituiert ist. Während die Protonen an C-3 Signale bei sehr hohem Feld zeigen ($dd(br)$ 0.65 und dd 0.75), liegt das Signal des dritten Dreiringprotons, wie Doppelresonanz-Experimente zeigen, als dd bei 1.54. Dieses ist nur mit einer allylischen Stellung dieses Protons vereinbar. Nach Zusatz von Eu(fod) $_3$ werden die sonst zusammenfallenden Signale der allylischen Protonen getrennt, und man kann durch Entkopplung zeigen, daß neben der exocyclischen Doppelbindung eine

CH_2 -Gruppe steht, die mit einem CH_2 -Multipllett bei ca 1.20 koppelt. Die Zuordnung der übrigen Signale läßt sich ebenfalls durch Entkopplungen sichern. Auch das Massenspektrum ist gut mit der Konstitution **3** vereinbar. Besonders deutlich ist die Abspaltung der Seitenkette unter Bildung eines Fragmentes C_7H_9 zu beobachten. Offen bleibt jedoch die Konfiguration an C-2, C-4 und C-7. Der Grundkohlenwasserstoff ist das bereits bekannte Sesquisabinen [3].

EXPERIMENTELLES

100 g Wurzeln von *Arctotis grandis* Thunb. (angezogen aus Samen vom Botanischen Garten Chelsea, Herbar Nr. 77/965) wurden frisch zerkleinert und mit Et_2O /Petrol 1:2 bei Raumtemp. extrahiert. Der Extrakt ergab nach DC (Sigel, GF 254) 1 mg **2**, Spuren von **1** und 16 mg **3** (Et_2O /Petrol 1:10).

12-Acetoxy-sesquisabinen (**3**): Farbloses Öl. IR: OAc 1745, 1230; $\text{C}=\text{C}$ 3080, 1650, 870 cm^{-1} . MS: M^+ m/e 262.193

($\text{C}_{17}\text{H}_{26}\text{O}_2$) (6%), —AcOH 202 (8); 202 —Me 187 (9);  $^+$

93 (60); $\text{H}_3\text{C CO}^+$ 43 (100).

$$[\alpha]_{24}^{25} = \frac{589}{-36.9} \frac{578}{-38.5} \frac{546 \text{ nm}}{-43.6} (c = 1.57, \text{CHCl}_3).$$

LITERATUR

1. Samek, Z., Holub, M., Drozda, B. und Gretarczyk, H. (1977) *Coll. Czech. Chem. Commun.* **42**, 2217.
2. Bohlmann, F., Burkhardt, T. und Zdero, C. (1973) in *Naturally Occurring Acetylenes*, Academic Press, London.
3. Terhune, S. J., Hogg, J. W., Bronstein, A. C. und Lawrence, B. M. (1975) *Can. J. Chem.* **53**, 3285.

SENKIRKINE AND A NEW SESQUITERPENE GLYCOSIDE, ISOPETASOSIDE, FROM *PETASITES JAPONICUS*

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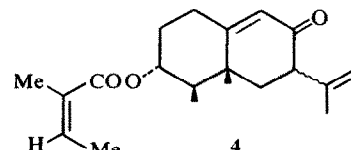
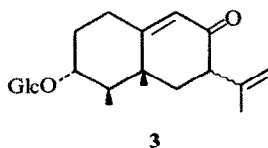
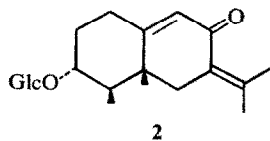
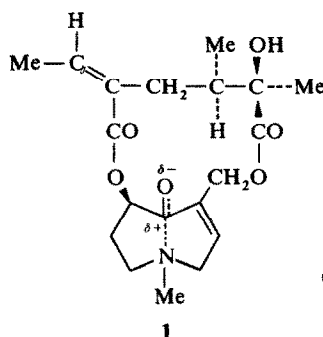
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Key Word Index—*Petasites japonicus*; Compositae; pyrrolizidine alkaloids; senkirkine; sesquiterpene glycoside; isopetasoside; isopetasol.

Petasites japonicus, which is used both as a food and herbal remedy, has been previously examined for sesquiterpenes [1–3], phenolics [4], and essential oils [5]. Carcinogenicity of the young flower stalks of this plant was reported by one of us (I.H.) [6] and in a search for the carcinogen(s), we isolated two new pyrrolizidine alkaloids, petasitenine and neopetasitenine, the structures of which have been elucidated [7, 8]. Furuya *et al.* [9] also isolated a pyrrolizidine alkaloid which they called fukinotoxin [9], but which in fact is petasitenine [10]. Petasitenine has recently been proved to be a carcinogen [11]. Further scrutiny of the alkaloids of *P. japonicus* has now resulted in the isolation of senkirkine

(**1**) [12], a minor alkaloid (the highest yield obtained: ca 0.0001% based on the weight of the dried plant material), which may also be responsible for the carcinogenicity of this plant. The amount of **1** varied with plant age and there was more in the rhizomes of the mature plant (e.g. 0.0026% based on dry wt.) than in the young flower stalks.

A new sesquiterpene glycoside, isopetasoside (**2**) was isolated and characterized as its tetraacetate. Based on spectral properties, the aglycone of **2** was deduced to be isopetasol [13–15]. Acid hydrolysis of **2** afforded glucose and isopetasol. Assignment of the β -anomeric configuration of the glucopyranoside moiety in **2** was based on the



PMR spectral evidence: a doublet due to the anomeric proton in the PMR spectrum of the acetate appeared at δ 4.62 with the coupling constant of 7.5 Hz. The D-configuration of glucose was determined by enzymatic means. Thus isopetasoside is (+)-isopetasol- β -D-glucopyranoside (2). It should be noted that isopetasoside (2) was isolated as an inseparable mixture together with a closely related glucoside (3), the presence of which was apparent from the PMR spectroscopy. On treatment with base (NaOMe-MeOH, room temp.) the mixture of 2 and 3 was converted to pure 2. There is thus the possibility that the double bond of the isopropenyl group in 3 isomerized in the course of extraction to produce 2; such ready isomerization is known to occur during the hydrolysis of petasin (4) to isopetasol [14].

EXPERIMENTAL

Mps were uncorrected. IR: JASCO Model IRS, CHCl₃. PMR: Varian HA-100 (100 MHz), δ -values, TMS as internal standard. MS: Hitachi RMU-6C, direct inlet system, 70 eV. Optical rotations: JASCO Model DIP-SL. For TLC, Si gel GF₂₅₄, PF₂₅₄ and alumina GF₂₅₄, PF₂₅₄-Type I (Merck) were used. Organic solns were dried over Na₂SO₄.

Isolation of senkirkine (1). Young flower stalks of *Petasites japonicus* were collected in late March and early April in Kawai, Niu, and Shokawa Villages of Gifu Prefecture (Japan), dried at room temp. and ground. The powdered material (53 kg) was extracted in EtOH at room temp. for 10 days. After filtration, the concd soln (ca 10 l) was acidified (pH 2) with 6N H₂SO₄, and the mixture was exhaust. extracted with Et₂O. The aq. phase was made basic (pH 8) with conc NH₄OH, and exhaustively extracted with CHCl₃. The CHCl₃ extracts were evapd giving a dark brown residue, which was chromatographed on Si gel (Merck, No. 7734) with MeOH. The alkaloidal fractions were concd, giving a yellow resinous material, which was chromatographed on Al₂O₃ (Merck, No. 1097) with CHCl₃. Further separation by PLC on Al₂O₃ with EtOAc gave senkirkine (1) (50 mg), mp 192–195° (EtOAc), $[\alpha]_D^{19}$ –11° (MeOH; c = 0.82), (lit. [12] mp 196.5–197.5°, $[\alpha]_D^{25}$ –16° (MeOH; c = 1.89)). (Found: C, 62.35; H, 7.35; N, 3.93. Calc. for C₁₉H₂₇O₆N: C, 62.45; H, 7.45; N, 3.83%). Identification with authentic 1 was performed by spectral (IR, PMR, MS) and TLC (Al₂O₃, EtOAc) comparison.

Isolation of isopetasoside (2). Dried and ground young flower stalks of *P. japonicus* (200 g) were extracted in EtOH (2 litre) at 50° for 7 hr. After filtration, the filtrate was concd. and the residue diluted with H₂O (100 ml) and washed with Et₂O (3 $\times$ 100 ml). The aq. phase was extracted with CHCl₃ (3 $\times$ 150 ml) and the combined extracts concd. The oily residue (370 mg) was purified by PLC on Si gel (MeOH-CHCl₃, 1:4) to give a mixture of 2 and 3 (72 mg) as a pale yellow oil. A mixture of 2 and 3 (35 mg) was acetylated with Ac₂O-Py, and the product purified by PLC on Si gel (EtOAc-CHCl₃, 1:2) to give a crystalline mixture of the two acetates (25 mg) in the ratio of 6:4. To a soln of the two acetates (14 mg) in MeOH (0.5 ml) was added a soln (1.5 ml) of NaOMe in MeOH (Na (100 mg) and MeOH (10 ml))

and the mixture stirred at room temp. for 40 min, neutralized by ion exchange resin (Amberlite CG-50), and filtered. The concd filtrate was purified by PLC on Si gel (EtOAc-CHCl₃, 1:1) yielding pure 2 (6.5 mg) as a colourless oil; yield 0.012%; $[\alpha]_D^{20}$ +7.3° (EtOH; c = 2.2); IR $\nu_{\max}$ cm⁻¹: 3375, 1652, 1624, 1610 (sh); PMR (CDCl₃) 0.98 (3H, s), 1.06 (3H, d, J = 6.0 Hz), 1.84 (3H, s), 2.09 (3H, s), 5.76 (1H, s); MS m/e : 396 (M⁺), 235, 234.

Isopetasoside tetraacetate. Isopetasoside (2) (6.5 mg) was acetylated with Ac₂O-Py and the product purified by PLC on Si gel (EtOAc-CHCl₃, 1:1) to afford the acetate (7.4 mg), mp 186–188° (n-hexane-C₆H₆); $[\alpha]_D^{20}$ –10.6° (CHCl₃; c = 1.2); IR $\nu_{\max}$ cm⁻¹: 1753, 1652, 1625, 1610; PMR (CDCl₃) 0.98 (3H, s), 1.04 (3H, d, J = 6.0 Hz), 1.85 (3H, br s), 2.01 (3H, s), 2.03 (3H $\times$ 3, br s), 2.08 (3H $\times$ 2, br s), 3.58 (1H, m), 3.70 (1H, m), 4.27 (2H, m), 4.62 (1H, d, J = 7.5 Hz), 4.85–5.40 (3H, complex pattern), 5.76 (1H, d, J = 1.5 Hz); MS m/e : 564 (M⁺). (Found: C, 61.78, H, 7.12. C₂₉H₄₀O₁₁ requires: C, 61.69; H, 7.14%).

Acid hydrolysis of isopetasoside (2). A soln of 2 (13 mg) in MeOH (1 ml) and 2.5 N HCl (1 ml) was gently refluxed for 90 min, concd and exhaustively extracted with CHCl₃. The combined concd CHCl₃ extracts were purified by PLC on Si gel (EtOAc-CHCl₃, 1:1) yielding isopetasol (4.5 mg), mp 124–125° (n-hexane-C₆H₆) (authentic sample, mp 124–125°; mixed mp 124–125°); $[\alpha]_D^{25}$ +72.5° (EtOH; c = 0.4) (lit [14], $[\alpha]_D$ +116.6° (CHCl₃; c = 1.04)). Identification was confirmed by IR, PMR, and MS comparison with an authentic marker. The presence of glucose in the aq. phase was proved by PC.

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REFERENCES

- Naya, K., Hayashi, M., Takagi, I., Nakamura, S. and Kobayashi, M. (1972) *Bull. Chem. Soc. Jpn* **45**, 3673 and references therein.
- Novotný, L., Kotva, K., Toman, J. and Herout, V. (1972) *Phytochemistry* **11**, 2795.
- Shirahata, K., Kato, T., Kitahara, Y. and Abe, N. (1969) *Tetrahedron* **24**, 3179.
- Sakamura, S., Yoshihara, T. and Toyoda, K. (1973) *Agric. Biol. Chem. (Tokyo)* **37**, 1915.
- Kikuchi, M. (1973) *Yakugaku Zasshi* **93**, 123.
- Hirono, I., Shimizu, M., Fushimi, K., Mori, H. and Kato, K. (1973) *Gann* **64**, 527.
- Yamada, K., Tatematsu, H., Suzuki, M., Hirata, Y., Haga, M. and Hirono, I. (1976) *Chem. Letters* 461.
- Yamada, K., Tatematsu, H., Hirata, Y., Haga, M. and Hirono, I. (1976) *Chem. Letters* 1123.
- Furuya, T., Hikichi, M. and Itaka, Y. (1976) *Chem. Pharm. Bull. (Tokyo)* **24**, 1120.
- Furuya, T. personal communication.
- Hirono, I., Mori, H., Yamada, K., Hirata, Y., Haga, M., Tatematsu, H. and Kanie, S. (1977) *J. Natl. Cancer Inst.* **58**, 1155.
- Briggs, L. H., Cambie, R. C., Candy, B. J., O'Donovan, G. M., Russell, R. H. and Seelye, R. N. (1965) *J. Chem. Soc.* 2492.
- Aebi, A. and Djerassi, C. (1959) *Helv. Chim. Acta* **42**, 1785.
- Herbst, D. and Djerassi, C. (1960) *J. Am. Chem. Soc.* **82**, 4337.
- Naya, K. and Takagi, I. (1968) *Tetrahedron Letters* 629.